Nuclear Reactions

RELEVANCE: Basic Research – reaction mechanisms

structure

exotic nuclei

analytical tool — neutron activation analysis astrophysics — origin of elements Applications:

space radiation effects – satellites, Mars landing materials science – T_c(high) superconductors

medical therapy — tumor treatment

nuclear power – 25% of U.S. electricity

DEFINITION:

A collision between two nuclei that produces a change in the nuclear composition and/or energy state of the colliding species; i.e., 2nd order kinetics

1. Nomenclature

A. Constituents

- 1. Projectile: nucleus that is accelerated (v > 0); $0 \rightarrow 2 \times 10^{12} \text{ eV}$
 - a. neutrons reactors
 - b. light ions $-A \le 4$: AGS (NY), FNAL (IL)
 - c. heavy ions A > 4 \rightarrow U: ANL (IL), MSU (MI), TAMU (TX), RHIC (NY)
 - d. electrons (and photons) SLAC (CA), CEBAF (VA), MIT (MA)
 - e. exotic beams: π , K, \overline{p} : FNAL
 - f. radioactive beams: MSU, ANL (IL) HHJRF (TN)
- 2. Target: fixed (v = 0), usually

But now have colliding beams, RHIC, FNAL

3. Products

ANYTHING PERMITTED BY CONSERVATION LAWS.

Note: since projectile kinetic energy can be converted into mass, only limit on mass-energy is beam energy.

B. Notation:

Target [projectile, light product(s)] heavy product

$$\left\{\begin{array}{cccccc} e.g., \ _{3}^{7}Li \ + \ _{82}^{208}Pb & & \\ & & \\ & & \\ & & \\ & & \\ \end{array}^{212}At + 3n & : & & \\ & & \\ & & \\ ^{208}Pb(^{7}Li, 3n)^{^{212}}At & \\ & & \\ & & \\ & & \\ \end{array}^{208}Pb(^{7}Li, 18C)^{^{197}}Au & \\ & & \\ \end{array}\right\}$$

C. Energetics: Q-values revisited

- 1. $Q = \sum \Delta(\text{reactants}) \sum \Delta(\text{products}) = \text{available energy}$
- 2. Example:

Pb(
7
Li, γ) reaction:
$$Q = \Delta(^{208}\text{Pb}) + \Delta(^{7}\text{Li}) - \Delta(^{215}\text{At}) - \Delta(\gamma,)$$
$$Q = -21.759 + 14.908 - (-1.263) - 0$$

$$Q = -5.589 \text{ MeV}$$
 ENDOTHERMIC

Accelerate ⁷Li to 5.589 MeV, no reaction. WHY?

II. Energetic Conditions for Reaction

- A. Energetic Threshold: E_{th} & Excitation Energy: E*
 - 1. <u>DEFINITION</u>: $\begin{cases} E_{th} \text{ is the minimum projectile energy necessary} \\ \text{to satisfy mass-energy and momentum conservation} \\ \text{(i.e., compensate for } -Q) \end{cases}$
 - 2. Derivation:
- a. Apply conservation Laws
 projectile + target → composite nucleus → products

$$E = \frac{p^2}{2m}; p = \sqrt[2]{2mE}$$

OR
$$M_p E_p = M_{CN} E_{CN}$$

$$\approx A_p E_p = A_{CN} E_{CN}$$

b. Combining

$$\begin{split} E_p + \Delta_p + \Delta_t &= \Delta_{CN} + (A_p/A_{CN})E_p + E^* \\ E_p - (A_p/A_{CN})E_p &= \Delta_{CN} - \Delta_p - \Delta_T + E^* \\ E_p &(1 - A_p/A_{CN}) &= E^* - [\Delta p + \Delta_T - \Delta_{CN}] \\ E_p &\frac{A_{CN} - A_p}{A_{CN}} &= E^* - Q \end{split}$$

$$E_p (A_T/A_{CN}) = E^* - Q$$

• IF $E^* = 0$ (minimum), then $E_p = E_{th}$ and

$$E_{th} = (A_{CN}/A_T)(-Q)$$

 $\bullet \ \ IF \quad E^* > E_{th} \quad \boxed{ E^* \ = \ (A_T\!/A_{CN}) \ E_p + Q }$

c. Example:

208
Pb(7 Li, γ) 215 At ; Q = -5.589 MeV
 $E_{th} = -Q \left(\frac{A_{CN}}{A_{T}} \right) = -(-5.589 \text{ MeV})(215/108) = 5.777 \text{ MeV}$

$$E_{CN} = E_{th} - E_p = 5.777$$
 -5.589 = 0.188 MeV

i.e., of total of 5.777 MeV, 5.589 MeV goes into mass ($M = E/c^2$) and 0.188 goes into kinetic energy of the composite nucleus.

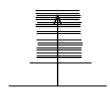
d. Example:
$$E_p = 40.0 \text{ MeV}$$
,

$$E^* = \left(\frac{208}{215}\right) 40.0 \text{ MeV} - 5.589 = 33.1 \text{ MeV}$$

• This energy is converted into heat

Fermi Gas model:
$$T \propto \sqrt{E^*/A}$$

 $\left\{1 \text{ MeV} \approx 10^{10} \text{ °K (kT)}\right\}$

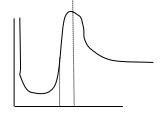


e. Accelerate to 5.777 MeV Still no reaction. WHY?

B. Coulomb Barrier - Charged Projectiles Only

1. Nucleus-Nucleus Charge Repulsion

$$\left\{ \begin{array}{ccc} V_{coul} = & \frac{(Z_p e)(Z_T e)}{d} & = & \frac{Z_p Z_T e^2}{R_p + R_T} \end{array} \right\}$$



2. <u>Potential Energy Surface</u>

Nuclear Reactions begin to occur when <u>tails</u> of nuclear matter distributions overlap



Bottom line: r_0 for reactions is greater than for potential well.

3. Net Result

$$\begin{cases} V_{\text{Coul}}^{\text{cm}} = \frac{1.44 \, Z_p Z_T \, \text{MeV} - \text{fm}}{r_0 (A_p^{1/3} + A_T^{1/3})} = \frac{0.90 \, Z_p Z_T}{(A_p^{1/3} + A_T^{1/3})} \, \text{MeV} \end{cases}$$

4. Momentum Conservation

CN must carry off some kinetic energy; same correction as for Mass-Energy Conservation (Q-value)

$$V_{Coul}^{lab} = \left(\frac{A_{CN}}{A_{T}}\right) V_{Coul}^{cm}$$

5. Example 208 Pb(7 Li, γ) 215 At

$$V_{Coul}^{lab} = \left(\frac{215}{208}\right) \frac{(82)(3)(0.9)}{(208^{1/3} + 7^{1/3})} = 29.1 \text{ MeV}$$
 NOW THINGS HAPPEN

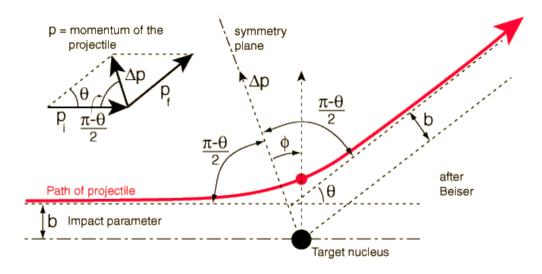
- 6. Diffuse nuclear surface fuzzes precision of Coulomb barrier
 - \therefore E_{th} is an **EXACT** condition

V_{Coul} is an APPROXIMATE condition

C. Centrifugal Barrier – NOT A MINIMUM CONDITION

Alpha Scattering Geometry

The <u>scattering</u> of the alpha particle by the central repulsive <u>Coulomb force</u> leads to a hyperbolic trajectory. From the scattering angle and momentum, one can calculate the impact parameter and closest approach to the target nucleus.



Rotational energy = $\frac{\ell(\ell+1)\hbar^2}{2I}$ where |l| = mvb and $I = \mu r^2$ where μ is the reduced mass of the system.

The rotational energy is not available for reaction!

D. Summary of Energetic Factors

- 1. $\underline{E}_p \ge \underline{E}_{th}$: Mass-Energy Conservation: ABSOLUTE CONDITION 1st law
- 2. $E_p \gtrsim V_{Coul}^{lab}$: Charge Repulsion Constraint BARRIER PENTRATION Probability low below this energy

- 3. $E_p \ge E_{rot}$: No constraint since $\ell = 0$ is always possible.
- 4. IN GENERAL

 $\left\{ \begin{array}{c} V_{\text{Coul}} > E_{\text{th}} \end{array} \right\} \text{; except for very light nuclei and neutral projectiles,} \\ \text{e.g., neutrons} \end{array}$

III. Reaction Probability: The Second-Order Rate Law

<u>Probability</u> $\equiv \underline{\sigma} = \underline{\text{cross section}}$

[SAME FOR CHEMICAL Rx]

A. Schematic Picture

1. ON-OFF nature of nuclear force suggests a simple geometric model:

Touching Spheres Model

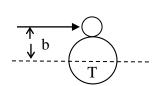
• <u>IF</u> projectile and target touch, REACTION;

$$b \leq R_p + R_T$$

• <u>IF</u> projectile and target don't touch, NO REACTION;

$$b > R_p + R_T$$

2. Bottom Line: Probability is proportional to cross-sectional area $Area = \pi (R_P + R_T)^2 = \sigma$



$$\sigma_{\rm R} = \pi r_0^2 (A_{\rm p}^{1/3} + A_{\rm T}^{1/3})^2$$

TOTAL REACTION CROSS SECTION

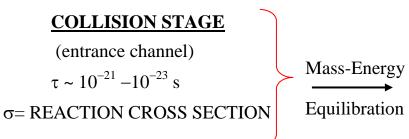
$$r_0 \approx 1.60 \text{ fm}$$

1 barn =
$$1 \times 10^{-24} \text{ cm}^2$$

$$= 1b$$

B. Definitions

1. **Sequential Process**



Collision Probability

DECAY STAGE

(exit channel)

$$\tau \sim 10^{-15} - 10^{-21} \,\mathrm{s}$$

Probability of forming a given product (many may be possible)

2.
$$\sigma_R = \sum \sigma(a,b)$$

i.e., Sum of all possible production σ 's equals the total reaction σ

3. Example
$$40\text{-MeV}$$
 $^4\text{He} + ^{232}\text{Th} \rightarrow ^{236}\text{U*} \rightarrow \text{products}$

C. Cross Section Measurements: Nuclear Reaction Rates

1. Review of Biomolecular Rate Law

$$A + B \rightarrow C + D$$
 [Elementary Reaction – only kind for nuclei]

$$\left\{ \text{Rate} = -\frac{d [A]}{dt} = k [A][B] \right\}$$

i.e., a second-order rate process

a. [A][B] factor: Collision Probability

Defines collision geometry; e.g., two gases or two liquids, molecular beam + gas, etc

b. k = rate constant = probability of reaction**IF**collision occurs.

$$k = f (\Delta H_a, T, structure, etc.)$$

2. Nuclear Case: Charged-Particle-Induced Reactions

$$k=\sigma$$
 ; [A][B] = $n_p n_T$ (projectile nuclei × target nuclei); definition of n is geometry dependent

$$R = \underline{Rate} = \underline{Number of Product nuclei/time} \ \sigma(a, b) \ \underline{\mathbf{OR}}$$

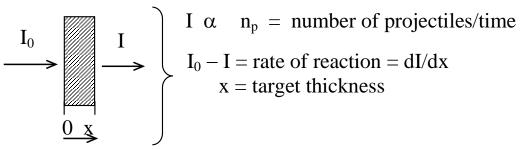
Number of Reactions/time

 σ_{total}

 $\sigma = cross section - must be given$

3. Thick Target: General Case

For thick targets, nuclear reactions remove a significant fraction of the beam:



 \therefore R = $(-dI/dx) = \sigma n_t I dx$, where $n_t = \rho N_0/(g-at. wt.)$ i.e., number density (N/cm³)

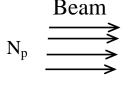
$$\int_{I_0}^{I} \frac{dI}{I} = -\sigma n_t \int_{0}^{x} dx$$

 $I = I_0 e^{-n\sigma x}$ Beer-Lambert Law

4. Thin Target Case

GEOMETRY:

Beam + infinitely thin target [thin = no shadowed nuclei]



fixed (i.e., v = 0); Area of target > Area of beam

Thin target Result:

Rate =
$$I\sigma(n_t x)$$

a.

$${}^{A}_{Z}X^{+q}$$
 \longrightarrow ${}^{A}_{Z}X^{+Z}$ \longleftarrow e^{-} (I electrons/projectile nucleus)

Projectile Beam

Target

Faraday Cup

To count projectiles, measure current i collected in Faraday cup

I = i / particle charge q

I is measured in Amps, since charged particles

 $q = Z_p e$ (ion charge)

e.g., 10nA ¹²C beam (stripped of electrons)

$$1nA = 6.28x10^9 e^- / s$$

$$10nA = 6.28x10^{10}e^{-}/s$$

However, each ¹²C has a charge of +6.

$$\therefore \#^{12}\text{C/s} = \frac{6.28x10^{10} e^{-} / s}{6e^{-}/^{12}C} = 1.05x10^{10} \ ^{12}\text{C/s} = \mathbf{I}$$

b. $(n_t x) = \#t \arg etatoms / cm^2 = \text{density } x \text{ thickness}$

c. TOTAL RATE R = Number of Reactions/unit time

$$R = I\sigma(n_t x)$$

5. Problem: What is the production rate of $^{266}_{106}$ Sg if a 100 µg/cm² target of $^{248}_{96}$ Cm is bombarded with a 1.0 µA beam of 22 Ne ions? σ (22 Ne, 4n) = 1.0 nb

$$R = I\sigma(n_t x)$$

$$\sigma = (1.0 \times 10^{-24} \text{ cm}^2)(10^{-9}) = 1.0 \times 10^{-33} \text{ cm}^2$$

$$(n_t x) = \underbrace{(100 \times 10^{-6})g (6.02 \times 10^{23} \text{ atoms/mole})}_{248 \text{ g/mole}} = \underbrace{2.43 \times 10^{17} \text{ atoms/cm}^2}_{cm^2}$$

$$I = 1.0 \ \mu A \times 6.28 \times 10^{12} \ e^{-}/\mu A \ / \ (10 \ e^{-}/Ne) = 6.28 \times 10^{11} Ne/s$$

$$R = (1.0 \times 10^{-33} \text{ cm}^2) \left(\frac{2.43 \times 10^{17}}{\text{cm}^2} \right) (6.28 \times 10^{11}/\text{s}).$$

$$R = 1.5 \times 10^{-4}/s = 0.54/hr$$

$$\sigma$$
 (²²Ne, f) = 2.5 b

$$R = 7.6 \times 10^5 / s$$

NOTE: 2 fragments/fission

i.e., humongous fission background