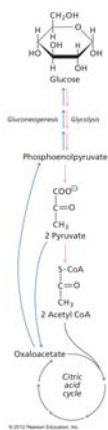
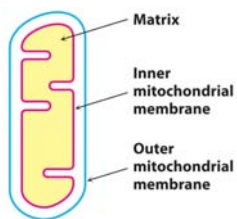


Preparation for Citric Acid Cycle

Chapter 18
Stryer Short Course

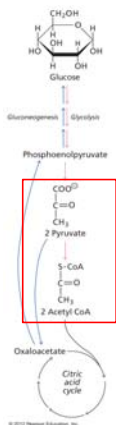
Overview

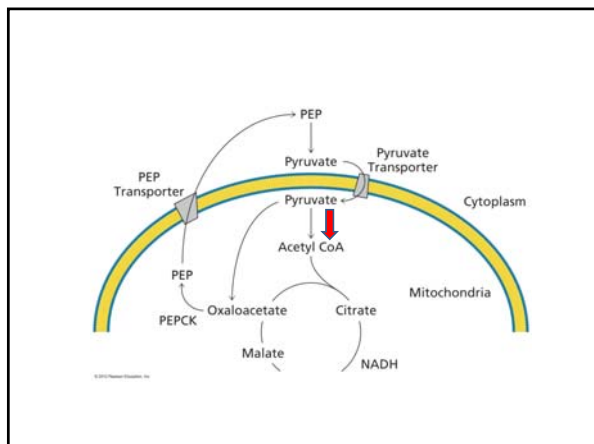
- Compartmentalization
 - Glycolysis: Cytosol
 - Citric Acid Cycle: mitochondria



Bridge

- Glycolysis
- Pyruvate dehydrogenase complex
 - Commitment of carbon away from carbohydrates
- Citric acid cycle





The Preparation Reaction

$$\begin{array}{c} \text{COO}^- \\ | \\ \text{C}=\text{O} \\ | \\ \text{CH}_3 \end{array} + \text{HS-CoA} + \text{NAD}^+ \xrightarrow{\text{Pyruvate dehydrogenase}} \begin{array}{c} \text{S-CoA} \\ | \\ \text{C}=\text{O} \\ | \\ \text{CH}_3 \end{array} + \text{CO}_2 + \text{NADH}$$

Pyruvate Acetyl CoA

Pyruvate Dehydrogenase Complex

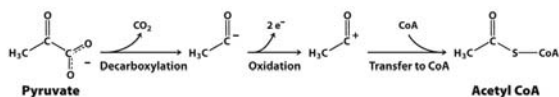
- Three distinct enzymes—in a massive complex
- Five chemical steps

Enzyme	Abbreviation	Number of chains	Prosthetic group	Reaction catalyzed
Pyruvate dehydrogenase component	E_1	24	TPP	Oxidative decarboxylation of pyruvate
Dihydrolipooyl transacylase	E_2	24	Lipoamide	Transfer of acetyl group to CoA
Dihydrolipooyl dehydrogenase	E_3	12	FAD	Regeneration of the oxidized form of lipoamide

Abbreviations: TPP, thiamine pyrophosphate; FAD, flavin adenine dinucleotide.

Formal Reactions

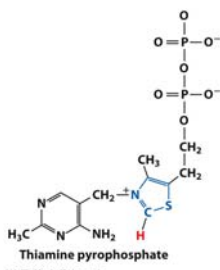
- Where are the problems in this formal reaction?
- Cofactors used to overcome problems!

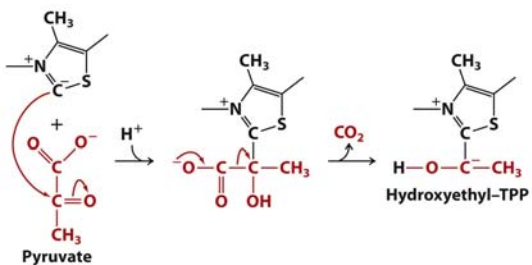


Unnumbered 18 p334e
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Pyruvate Dehydrogenase (E₁)

- TPP cofactor:
Decarboxylation of α carboxyketones
- Stabilization of "acyl anion"
- Draw mechanism of decarboxylation

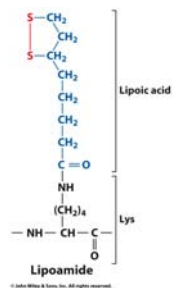


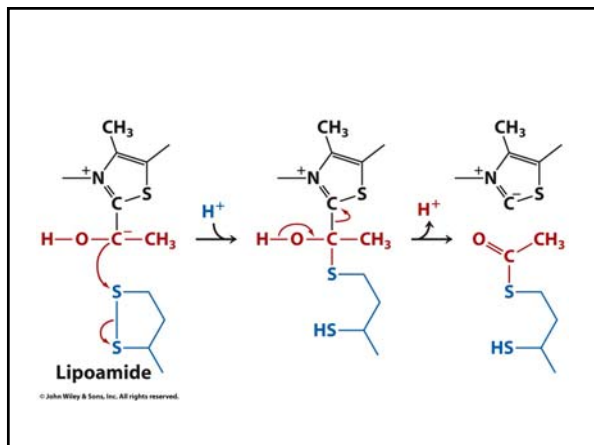


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Dihydrolipoamide Acyltransferase (E₂)

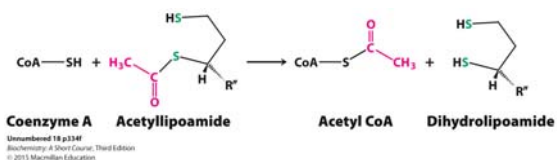
- Transfer catalyzed by E1
- Serves as a linker to “swing” substrate through subunits
- Mechanism of redox





Step 3: transfer

- Maintenance of high energy bond
- Acetyl CoA product is made
- Lipoamide still reduced—not catalytically viable at this point



Dihydrolipoamide dehydrogenase (E₃)

- Redox of prosthetic FAD/FADH₂
- Still not a regenerated catalyst!

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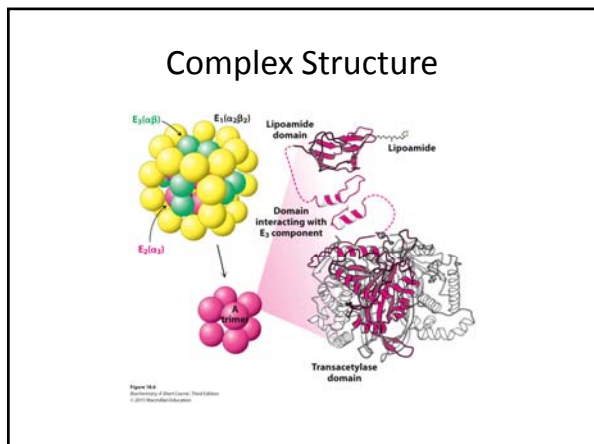
Step 5: NADH produced

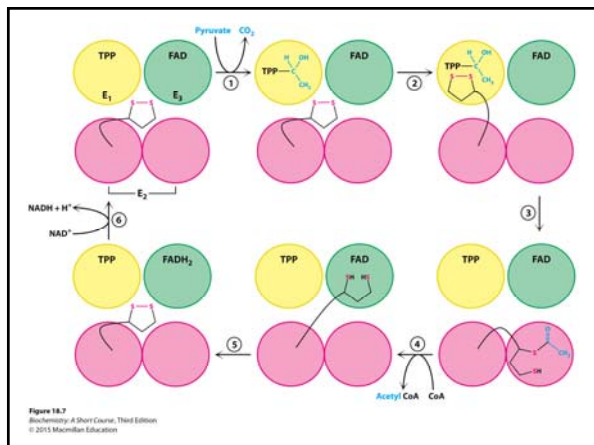
- Catalytic prosthetic group is restored by stoichiometric NAD+
- Overall redox:

Dihydrolipoamide + FAD → Lipoamide + FADH₂ → FAD + NADH + H⁺

Oxidation energy of one carbon atom used to produce high energy thioester and NADH

Green: Reactants
Purple: products
Red: catalysts





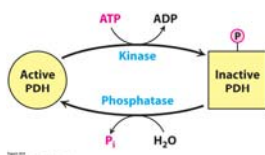
Regulation

- Metabolically irreversible
- Local regulation conserves glucose
- Pyruvate Dehydrogenase Complex inhibited by product
 - NADH
 - Acetyl CoA
- When these are at high concentration in mitochondria, pyruvate is not wasted

Figure 18.8
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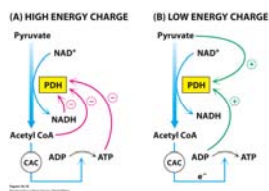
Covalent Regulation

- Kinase and phosphatase activity built into core of PDH
 - Kinase inactivates PDH
 - Phosphatase activates PDH
- Activity controlled by
 - Local energy charge
 - Hormones



Energy Charge Regulation

- High energy charge activates PDH kinase
- Low energy charge activates PDH phosphatase



Hormone Regulation

- | | |
|--|--|
| <ul style="list-style-type: none"> • α-Adrenergic receptor <ul style="list-style-type: none"> – Liver – Phosphoinositol pathway – Activates PDH phosphatase – More acetyl CoA – Purpose: Energy | <ul style="list-style-type: none"> • Insulin <ul style="list-style-type: none"> – Liver, adipose – Insulin receptor – Activates PDH phosphatase – More acetyl CoA – Purpose: store as fatty acids |
|--|--|

Related Pathologies

- PDH can be inactivated by
 - Genetic defect of PDH phosphatase
 - cancer enhancement of PDH kinase
 - Beriberi (TPP deficiency)
 - Arsenic (binds lipamide)
- Results in
 - No entrance from glucose into citric acid cycle
 - Glucose metabolized anaerobically
 - Especially bad for nervous system, which can only make acetyl CoA from glucose (others can use fats to make acetyl CoA)
